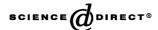
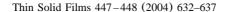


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Investigation of the electrical properties and reliability of amorphous SiCN

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Abstract

Amorphous SiCN (a-SiCN) is a candidate for barrier dielectric and has lower dielectric constant ($k \sim 5$) relative to the commonly used barrier SiN ($k \sim 7$). In this study, we investigate the leaky behavior and barrier characteristics of a-SiCN ($k \sim 5$) doped with different nitrogen concentration. The leaky behavior of a-SiCN is Poole–Frenkel at high electric field. This is similar to SiN. Thermal stability of a-SiCN films is good enough to meet the prescription in back-end-of-line fabrication process. The bias-temperature stress (BTS) test has been conducted on a-SiCN to investigate the barrier ability against copper $\frac{15}{16}$ penetration. We find that a-SiCN could sustain the stress of electric field up to 4 mV/cm at 150 °C. In addition, a-SiCN films with higher nitrogen concentration exhibit better barrier property against copper penetration. The SIMS spectra also have been used to monitor the distributions of copper after different BTS conditions, confirming our inference on leakage mechanism.

Keywords: Barrier dielectric; a-SiC; BTS; Cu penetration; Low-k

1. Introduction

Copper wiring is currently used in ultra large scale integrated circuits to reduce the signal delay [1]. Compared to Al wiring, Cu has better resistivity toward electromigration. However, one of the reliabilities in Cu metallization is dielectric degradation caused by Cu ion drift. Copper could rapidly drift in silica-based low-*k* during bias-temperature stressing [2]. Therefore, it is necessary to introduce a barrier dielectric between Cu wiring and dielectric insulators to prevent the Cu ion drift/diffusion [3,4]. The typically used barrier dielectric is SiN, which could efficiently prevent the Cu drift [5]. However, the dielectric constant of SiN is lager (approx. 7). To reduce the effective dielectric constant of intrametal and intralayer dielectrics, SiN must be substituted

[6]. Studies on barrier dielectrics which have lower dielectric constants related to SiN would be attractive [7,8]. In this study, we demonstrate the SiC-based dielectric films with various nitrogen concentrations. Electrical characteristics and material analysis will be used to explore leakage behaviors of barrier dielectrics.

2. Experimental

The silicon carbide films with various concentrations of nitrogen were deposited with tri-methyl-silane source using plasma-enhance chemical vapor deposition system. The deposition temperature was 350 °C. Various concentrations of nitrogen in the films were obtained by controlling by NH $_3$ gas flow rate during deposition. The pressure in the chamber is kept at 3 Torr during the deposition. The silicon carbide film was deposited on p-type silicon wafer with a resistivity of 15–25 Ω cm. The code name of the pure silicon carbide film in this

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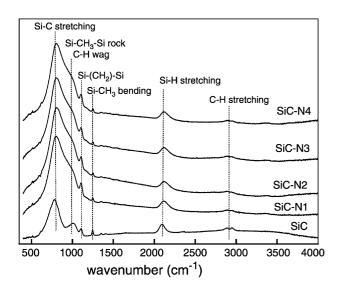


Fig. 1. FTIR spectrum of silicon carbide with or without NH_3 gas flow during deposition process.

study is 'SiC'; besides the code names of silicon carbide films with various flow rate of NH₃ gas are 'SiC-N1', 'SiC-N2', 'SiC-N3' and 'SiC-N4'. The flow rate of NH₃ gas for SiC-N1, SiC-N2, SiC-N3 and SiC-N4 were increasing from 250 to 500 sccm. X-Ray photoelectron spectra (XPS) were recorded by a VG Escalab MKII spectrometer using Al Kα (1253.6 eV) radiation for measuring the composition of silicon carbide films. The infrared spectrometry was performed from 4000 to 400 cm⁻¹ using a Fourier transform infrared (FTIR) spectrometer calibrated to an unprocessed bare wafer, for determining the chemical structure of the silicon carbide film. The refractive index and thickness of the silicon carbide with or without nitrogen were measured with an n and k analyzer and SORRA ellipsometer. Metal-insulator-semiconductor (MIS) structures of Cu/ silicon carbide/silicon as well as Al/silicon carbide/ silicon were used to investigate the behavior of basic electrical characteristics.

The dielectric constants and leakage current of the SiC films were investigated using capacitance-voltage (C-V) and current-voltage (I-V) characteristics measurement on MIS capacitor structure. A Keithley Model 82 CV meter at 1 MHz was used to measure the dielectric constant of the film with a gate electrode area of 0.0053 cm². The current-voltage (I-V) characteristics of SiC films were measured by an HP4156. The gate-electrode area of the MIS structure was 0.00133 cm² for I-V measurement.

The completed Al and Cu gated MIS capacitors were then bias-temperature-stressed (BTS) at 150 °C with gate electrode biases ($V_{\rm gate}$). The stress time was 1000 s, which is long enough for the mobile charges to drift

across the stacked insulator layer. The bias was provided by an HP4156B semiconductor parameter analyzer.

3. Results and discussions

Fig. 1 shows FTIR spectrum of silicon carbide films with and without the various flow rate of NH₃ gas during the deposition process, and the important regions of silicon carbide film were labeled in the figure and indicated as follows. The peak at 780 cm⁻¹ is identified as Si–C stretching bond. In addition, Si–CH₂–Si rock and C–H wag peak are near 990 cm⁻¹, C–F bending peak is at 1100 cm⁻¹, Si–CH₃ bending peak is at 1245 cm⁻¹, Si–H stretching peak is near 2100 cm⁻¹ and C–H stretching peak is near 2960 cm⁻¹. All of these bonds make the surface of the silicon carbide film hydrophobic. From the figure, there were not any nitride derivatives or nitride in any of the samples, and it was obvious that SiC–N1 ~ SiC–N4 have stronger Si–C stretching bonds than the pure silicon carbide, SiC.

XPS spectra for all of the specimens are shown in Fig. 2, and the major peak for Si_{2p} (~ 100 eV), Si_{2s} (~ 155 eV) and C_{1s} (~ 285 eV) of every specimens are clearly evident in it. The peak of N_{1s} (~ 398 eV) appears in SiC–N1–SiC–N4, but was absent in pure silicon carbide SiC. The enlargements of N_{1s} for all the specimens are shown in Fig. 3. It certificates the existence of SiN bonding in SiCN samples. The peak heights of SiC–N1 \sim SiC–N4 are higher than that of SiC. On the contrary, the carbon contents of the sample SiC–N1 \sim SiC–N4 is smaller than that of SiC. Moreover, the content of silicon remains constant for all the specimens. The atom percentage of nitrogen content for SiC–N1, SiC–N2, SiC–N3 and SiC–N4 is 14.37%,

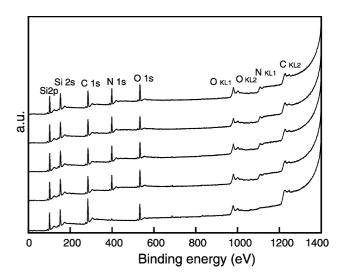


Fig. 2. XPS (X-ray photoelectron spectra) spectra for all the specimens.

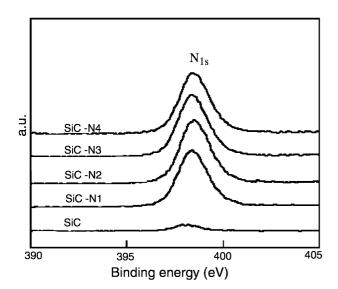


Fig. 3. Enlargement of XPS spectra of N_{1S} peaks in Fig. 2. The atom percentage of nitrogen content for SiC–N1, SiC–N2, SiC–N3 and SiC–N4 is 14.37%, 15.62%, 16.31% and 16.64%, respectively.

15.62%, 16.31% and 16.64%, respectively. Although the change is not significant, the nitrogen content certainly increases with increasing ammonia flow rate during the deposition.

The dielectric constant (k) values for all the specimens were demonstrated in Table 1. The pure silicon carbide film had a dielectric constant approximately 3.8 and the dielectric constants of nitrogen-containing films, SiCN, were higher from 4.3 to 4.5. The increase of dielectric constant is due to the appearance of SiN bonds in a-SiC films. The tetrahedral structure of SiC is distorted with additive of SiN bonds and then polarization chemical structure is raised [9].

The leakage current densities for all the films are sketched in Fig. 5. It shows clearly that the leakage current of pure silicon carbide (SiC) is much larger than the doped films. The samples we study are not crystallized SiC film but amorphous ones. Nitrogen is not an active dopant without activation process. On the contrary, a large amount of SiN bonds, which exist randomly in a-SiC films make the structure of a-SiC film more disorderly. In addition, the bonding energy of SiN is higher than SiC bond. The carrier would transport through the SiCN films difficultly. As a result, the leakage currents of a-SiCN films are smaller than a-SiC ones. The curves of SiC-N1-SiCN4 are almost similar so that we can put them on a par. Increasing the temperature increases the leakage current density. Furthermore, the J-E curves are symmetric both in positive and negatives bias, and the mechanism of leakage current is Poole-Frenkel (P-F) emission as the electrical field is above 2.5 mV/cm. Poole-Frenkel [10] emission is due to field-enhanced thermal excitation of

Table 1 The dielectric constant (k) values for all the specimens

	SiC	SiC-N1	SiC-N2	SiC-N3	SiC-N4
Dielectric constant(k)	3.83	4.50	4.47	4.34	4.40

trapped electrons in the insulator into the conduction band. The current density is given by:

$$J = J_0 \exp\left(\frac{\beta_{\rm PF} E^{1/2} - \phi_{\rm PF}}{k_{\rm B} T}\right)$$

where $\beta_{\rm PF} = (e^3/4\pi\varepsilon_0\varepsilon)^{1/2}$, $\phi_{\rm PF}$ is the contact potential barrier, $k_{\rm B}$ the Boltzmann constant, e the electronic charge, E the applied electric field, ε_0 the dielectric constant of free space, and ε the relative dielectric constant. Poole–Frenkel mechanism can be identified by comparing the theoretical value of $\beta_{\rm PF}$ with the experimental value of $\beta_{\rm exp}$ obtained by calculating the slope of the curve $\ln(J/E)-E^{1/2}$. Fig. 4 shows that the logarithm of leakage current of samples SiC–N4 is linearly related to the square root of the applied electric field, which is closed to Poole–Frenkel emission. The value of $\beta_{\rm exp} = 5.8 \times 10^{-23} \; ({\rm Jm}^{1/2}/{\rm V}^{1/2})$ which is close to the theoretical value of $\beta_{\rm PF} = 5.72 \times 10^{-23} \; ({\rm Jm}^{1/2}/{\rm V}^{1/2})$ as the dielectric constant is 4.4.

We investigated the impact of thermal annealing on the silicon carbide film. A silicon carbide film, SiC-N4 was annealed at temperature from 400 to 550 °C for more than 1 h in N_2 ambient. Fig. 6 shows the J-E curves of SiC-N4 with Cu gate before and after thermal

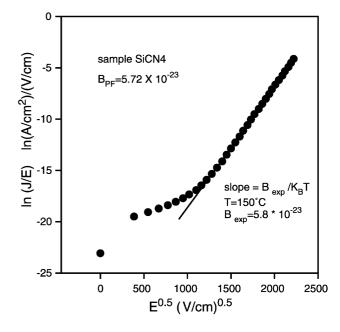


Fig. 4. J-E curve on a log (J/E) vs. $E^{1/2}$ plot of sample SiC-N4 showing Poole-Fenkel conduction at high electric field region.

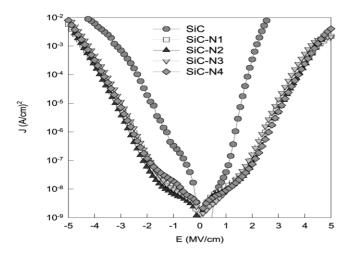


Fig. 5. *J–E* curves for all the samples. The area of electrodes we used is 0.00133 cm². The leakage current of SiCN based films are lower than the a-SiC films.

annealing. There is not much change in the leakage current after thermal annealing. Also, the FTIR spectrum does not vary after thermal treatment. Hence, the thermal treatment at temperature up to 550 °C does not have significant effect on the silicon carbide film. Consequently, the thermal stability of silicon carbide film is good enough to meet the prescription in back-end-of-line (BEOL) fabrication process.

The effect of NH₃ flow rate on the barrier ability of the prepared samples was also investigated, the flow rate of NH₃ used for the preparation of sample SiC–N4 being twice as high as the one used for preparing SiC–N1. Although the leaky behaviors of both samples are similar, as shown in Fig. 5, we performed the bias-

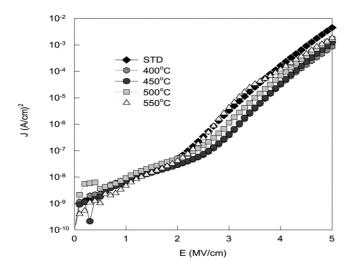


Fig. 6. *J–E* curves of SiC–N4 samples after thermal annealing at 400, 450, 500 and 550 °C for 1 h. Dielectric properties of SiC–N4 film remain stable up to 550 °C.

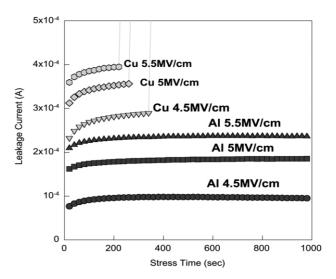


Fig. 7. The relationship of leakage current and stress time for SiCN1 films with Cu and Al gates during BTS test. BTS tests were conducted at 150 °C with electric fields of 4.5 mV and 5.5 mV/cm. Dielectric breakdown is easily occurred at Cu electrode samples at critical BTS condition (>4.5 mV/cm).

temperature stress measurement on them (Fig. 6). Biastemperature stressing (BTS) is a typical manner to estimate the barrier ability against the Cu penetration. We find that SiC–N1 and SiC–N4 exhibit excellent electrical stability even after BTS (3 mV/cm, 150 °C, 1000 s). Figs. 7 and 8 exhibit the $I_{\rm gate}-t_{\rm stress}$ curves of SiC–N1 and SiC–N4 with either Al or Cu gate during BTS measurement in high electric field. It is evident that the leakage current of SiC–N1 was almost one order of magnitude larger than that of SiC–N4. It is

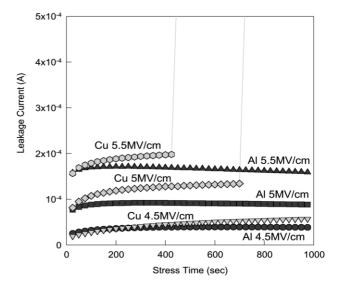


Fig. 8. Leakage current-stress time curves of SiC-N4 with Cu/Al gates during BTS at 150 °C.

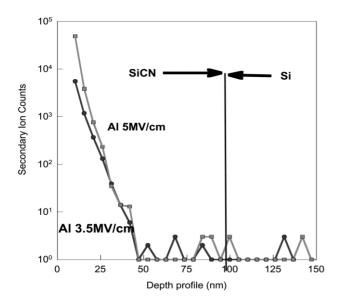


Fig. 9. SIMS depth profile of SiCN4 with Al gate after BTS measurement $(3.5/5 \text{ mV/cm}, 150 ^{\circ}\text{C}, 1000 \text{ s})$. The counts of Al remain at the same magnitude even after BTS with 5 mV/cm for 1000 s at 150 $^{\circ}\text{C}$.

worthy to mention that the leakage current is lower and becomes steady in the electric field below 3.50 mV/cm with increasing stress periods. However, it goes up for electric field greater than 3.75 mV/cm as shown in Figs. 7 and 8. The difference between Cu and Al curves is mainly due to the Cu penetration [11]. When Cu ions were drifted in high electrical field at high temperature, the moving Cu ions would produce the ion current. In addition, the Cu ions are regarded as trap centers in dielectrics, which could assist the carrier to transport

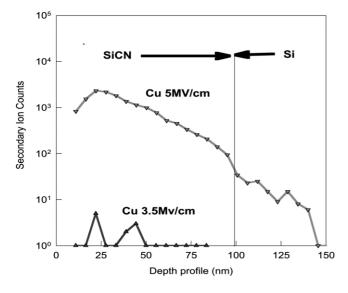


Fig. 10. SIMS depth profile of SiCN4 with Cu gate after BTS measurement (3.5/5 mV/cm, 150 $^{\circ}$ C, 1000 s). The Cu ions obviously penetrate through the SiCN4 film after BTS with 5 mV/cm.

[12]. The leakage currents increase abruptly at 4.5~mV/cm for SiC–N1 and 5.0~mV/cm for SiC–N4 indicating breakdown.

The difference between the increasing and decreasing behaviors of leakage currents can be explained by material analysis. Figs. 9 and 10 illustrate the secondary ions mass spectra (SIMS) of SiC–N4 with either Cu or Al electrode after BTS measurement in the condition of 3.5 and 5.0 mV/cm for 1000 s at 150 °C. It is clear that the copper counts in the SiC–N4 layer of the MIS structure after BTS with 5 mV/cm are much larger than that after BTS with 3.5 mV/cm, which almost remains zero. Furthermore, the Al counts remain the same in the silicon carbide layer of the MIS structure after 3.5 mV/cm BTS and 5.0 mV/cm BTS.

4. Conclusions

We have investigated the leaky behavior and barrier characteristics of a-SiCN containing different nitrogen concentrations. The leaky behavior of a-SiCN is Pool–Frenkel in high electric field region. Thermal stabilities of a-SiCN films are good enough to meet the prescription in BOEL fabrication process. We performed the bias-temperature stress (BTS) on a-SiCN films to test the barrier ability against Cu penetration. Experimental results indicate that a-SiCN films containing higher nitrogen concentration exhibited better barrier ability. The dielectric breakdown is due to the penetration of Cu.

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