



Bioconversion with whole cell penicillin acylase in aqueous two-phase systems

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Abstract

The deacylation of Pen G was carried out by using recombinant *E. coli* in an aqueous two-phase system consisting of polyethylene glycol and potassium phosphate solution, which partitions the cells to the bottom phase and the products to the top phase. Bioconversion and product separation were carried out in the same reactor. Repeated batch conversion was employed ten times and enzymic activity showed only a slight decline. When pure enzyme was used for bioconversion in an aqueous two-phase system, the decline was fast and bioconversion using whole cell penicillin acylase was better than that obtained using the pure acylase. © 1999 Elsevier Science Ireland Ltd. All rights reserved.

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1. Introduction

Enzymic hydrolysis of penicillin G is commonly employed in the industrial production of 6-aminopenicillanic acid (6-APA) for semisynthetic penicillins. Since the enzyme is expensive, reuse of enzyme is always required. There are three main approaches for repeated use of penicillin acylase. One is immobilization of the enzyme to a solid support [1–3], the second is retention of the enzyme in a membrane reactor [4,5], and the third is enzymic conversion in an aqueous two-phase system (ATPS) [6].

An ATPS is formed by adding two solutions of water-soluble polymers together or mixing a solution of water-soluble polymer with another salt solution. Since an ATPS has low interfacial tension between two phases, mass transfer can be easily carried out when the system is stirred. Furthermore, the partition characteristics of biocatalyst and product in an ATPS can be employed as an extractive bioconversion [7]. Andersson et al. made a review of bioconversions in aqueous two-phase systems [8].

In this study, an approach using whole cell penicillin acylase in an ATPS is developed. Most applications using whole cell are based on immobilization [9–11]. Rodriguez et al. [12] developed a two-phase dispersion system consisting of a suspension of cells and agar in oil. The proposed ATPS is composed of polyethylene glycol (PEG) and phosphate salt. The cells are partitioned to the bottom phase and the product to the top phase. Using whole cell in the ATPS has some advantages such as improving the partition of the biocatalyst to one phase and eliminating the enzyme purification step. Moreover, for repeated batch enzymic hydrolysis of penicillin G, little enzyme is lost when the top phase is removed. The operation of deacylation is prolonged by repeated use of the immobilized whole cell penicillin acylase.

2. Materials and methods

2.1. Materials

The potassium salt of penicillin G (Glaxo, UK) was used for conversion studies. The potassium salt of penicillin G, 6-APA and phenylacetic acid (PAA) were

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purchased from Sigma. Polyethylene glycol was PEG 6000 with the molecular weight between 5000 and 7000 (Merck, Germany). All chemicals were of reagent grade.

2.2. Microorganism and medium

A genetically engineered strain, *E. coli* ATCC 9637 (PGL5) [13] for producing penicillin acylase was used in this study. The microorganisms, which were provided by Union Chemical Laboratories (Hsinchu, Taiwan), had the property that penicillin acylase production was repressed by glucose and was not induced by phenylacetic acid addition. The microorganisms were maintained by monthly transfer on LB agar plates and stored at 4°C. Choramphenicol 30 µg ml⁻¹ was added to the media of the agar plates to maintain a selection pressure. The composition of the medium was as follows: sorbitol 40 g l⁻¹, yeast extract 5 g l⁻¹, KH₂PO₄ 5 g l⁻¹, Na₂HPO₄ 4 g l⁻¹, (NH₄)₂SO₄ 7 g l⁻¹, NH₄Cl 1.2 g l⁻¹, MgSO₄·7H₂O 2.4 g l⁻¹, CaCl₂ 0.02 g l⁻¹, and trace metal solution 10 ml l⁻¹. The pH value of the medium was 7.

2.3. The aqueous two-phase system

An ATPS consists of two separated aqueous phases. The principle of separation in the ATPS is the non-uniform distribution of substances between the two phases [14]. The partition coefficient *K* is defined as:

$$K = C_T/C_B$$

where *C_T* is the concentration of the substance in the top phase and *C_B* is the concentration of the same substance in the bottom phase.

Another important parameter of the ATPS is the phase volume ratio, *R*, which is defined as:

$$R = V_T/V_B$$

where *V_T* is the volume in the top phase and *V_B* is the volume in the bottom phase.

The ATPS in the present study was composed of PEG 6000 (15% w/w) and potassium phosphate solution (0.55 molarity). For determination of the partitions of Pen G, 6-APA, PAA, and cells, a 15-ml test tube was used. A vortex mixer was applied to mixing. After mixing of the ATPS, a batch-scale centrifuge was employed for separation at 1500 rpm for 2 min. Experiments on the bioconversion in the ATPS were carried out in a 500 ml flask with 15 g of the PEG 6000, 18 g of cells, and 0.55 M of the potassium phosphate solution. The working volume was 100 ml and incubation involved shaking at 150 rpm at 32°C for 60 min. After stopping the shaking, for 15 min, the system became two phases. Samples were carefully taken from both top and bottom phases with a pipette without disturbing the interface.

2.4. Assay

The concentration of Pen G was determined by using HPLC with an Interstil 7 ODS-3 4.6 × 250 nm (Verco-pak, Taiwan). The solvent system for the column was 20% acetonitrile, 0.1% triethylamine, 0.1% H₃PO₄ and 50 mM KH₂PO₄ at a flow rate of 1.0 ml min⁻¹. The detection was done by absorption at 220 nm with a UV detector. The concentrations of 6-APA and PAA are determined by using HPLC with an Interstil C₄ 5 µm column (GL Sciences, Japan). The solvent system for the column was 10 mol m⁻³ of (NH₄)₂HPO₄ and methanol (4:1 v/v) at a flow rate of 0.6 ml min⁻¹. The detection was done by absorption at 225 nm. Cell concentration was measured using a spectrophotometer at 600 nm and was correlated to the wet weight with a standard curve. Enzyme activity was determined according to the method proposed by Kornfeld [15].

3. Results and discussion

For bioconversion in an aqueous two-phase system, partition and stability of the enzymatic reaction are two important characteristic factors. The effects of composition of the ATPS and operating conditions on the partition and enzymic reaction are given below.

3.1. Partition of cells in the ATPS

Partition of cells to the bottom phase requires that the partition coefficient *K* should be very small. The experimental data show that for the ATPS the partition coefficient was less than 0.01 and the phase volume ratio was 4 at pH 7.8 and 32°C.

3.2. Partition of penicillin G, 6-APA and PAA

By applying an ATPS for bioconversion, the enzyme and products should be partitioned to different phases, especially for a system with product inhibition [16,17]. The partition coefficients of Pen G, 6-APA, and PAA in the ATPS under conditions of pH 7.8 and 32°C gave *K_{penG}* = 64, *K_{6-APA}* = 7.3 and *K_{PAA}* = 8.55. Hence, Pen G, 6-APA, and PAA were partitioned to the top phase. Although the partition coefficient of Pen G was high, most of the Pen G were deacylated to 6-APA during the period of bioconversion.

3.3. Effect of Pen G

Pen G had the effect of inhibiting, the activity of penicillin acylase [18,19]. A high concentration of Pen G in the ATPS decreased the conversion rate. Fig. 1 shows the relative activity with respect to the concentration of Pen G. The relative activity of 100% peni-

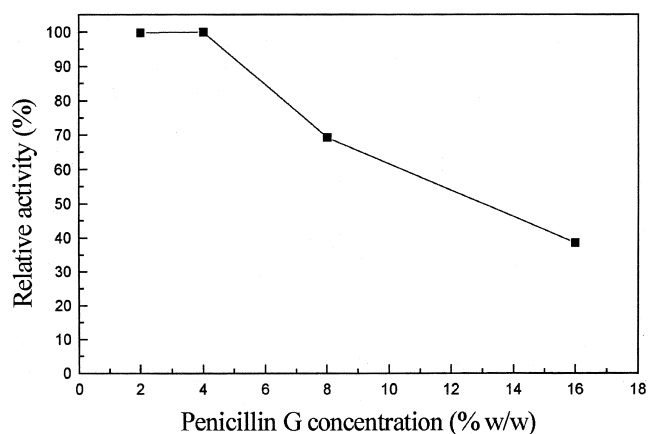


Fig. 1. Relative activity of whole cell penicillin acylase with respect to the penicillin G concentration at pH 7.8 and 32°C.

illin acylase was equal to 32 IU g⁻¹, wet weight. The concentration of Pen G up to 4% (w/w) did not have the effect of inhibition. The designated concentration of penicillin G for the reaction was 4% (w/w).

3.4. Effect of phosphate solution

Potassium phosphate in the ATPS had a substantial effect on the activity of penicillin acylase [20]. Fig. 2 shows the relative activity of whole cell penicillin acylase with respect to the concentration of potassium phosphate solution. Highest relative activity was obtained around 0.55 molarity.

3.5. Effect of temperature

Four different temperatures were employed to investigate the temperature effect on the bioconversion. Fig. 3 shows the relative activity with respect to temperature. A temperature above 32°C had almost the same

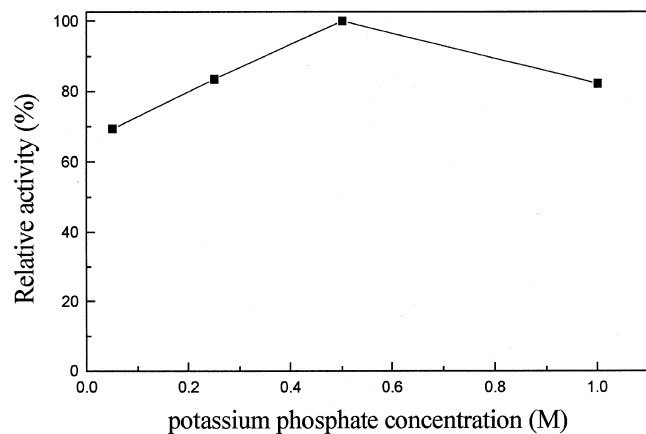


Fig. 2. Relative activity of whole cell penicillin acylase with respect to the concentration of the potassium phosphate solution at pH 7.8 and 32°C.

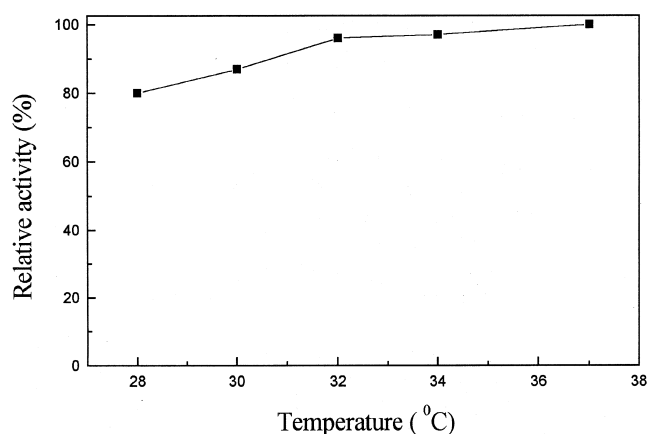


Fig. 3. Relative activity of whole cell penicillin acylase with respect to temperature at pH 7.8.

relative activity of bioconversion. Since at higher temperatures the production of penicillin acylase in the cells gave an inactive form [21], the designated temperature for reaction was 32°C.

3.6. Repeated batch conversion

Repeated batch operation was performed by removing the top phase and replacing it with a new top phase which had the same initial feed composition of the top phase of the ATPS with fresh penicillin G (4% w/w). Each batch operation took 60 min. Fig. 4 shows the experimental results of repeated operation. Penicillin G hydrolysis remained above 96% for ten repeated operations without adding fresh cells. The phase volume ratio of the ATPS was not significantly changed. The higher values of 6-APA concentration during the second to fourth repeated operations might be due to some 6-APA from the initial operation remaining in the bottom phase. The experimental results indicated that

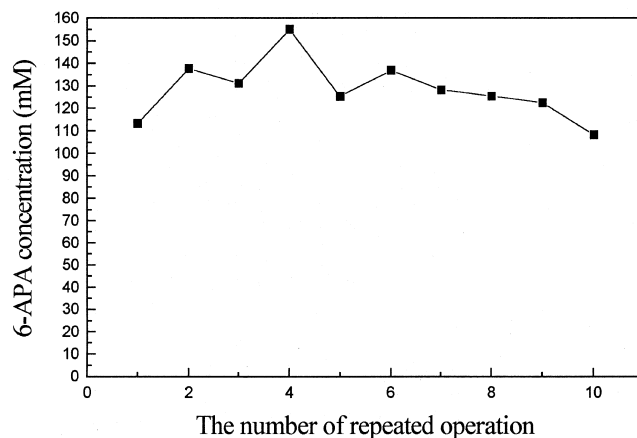


Fig. 4. Repeated batch conversion of penicillin G (4% w/w) to 6-APA (concentration in the top phase) with whole cell penicillin acylase at 32°C.

using a biocatalyst consisting of whole cells in the ATPS gave a better stable system. In a similar experiment [6] which was carried out using pure enzyme for deacylation of penicillin in the ATPS, fresh enzyme had to be added even for four repeated operations. Using the whole cell acylase significantly improves performance.

4. Conclusion

A whole cell penicillin acylase in an ATPS has been developed for bioconversion of penicillin G to 6-APA. Cells were partitioned to the bottom phase while the products, 6-APA and PAA, were partitioned to the top phase. Since the concentrations of penicillin G, potassium phosphate, and 6-APA could degrade the activity of penicillin acylase, an ATPS was employed to resolve the problem.

Using whole cell penicillin acylase for deacylation provided several significant advantages including:

1. Enzyme purification was not required.
2. Whole cell penicillin acylase could be re-used ten times before the enzymic activity began to decline.

In the present study, experiments were carried on a small scale. Although scale-up could be applied easily, details of the operation require further investigation.

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